

# Chemical Compounds



Our chemistry lab is in complete disarray! We need something to contain the mess. Cut out the element squares along the dashed lines and rearrange them, joining edges to form "chemical compounds". The chemical compounds read left-to-right or top-to-bottom across adjacent edges. Do not rotate the element squares; use the underlines to help prevent accidental rotations.

energy method <u>Ke</u> liquid nitrogen <u>T</u> boiling	tension <u>T</u> boiling	metal <u>Ow</u> critical prize <u>O</u> research	goggles <u>O</u> research	gas assistant <u>S</u> closed vapor <u>N</u> organic	table <u>N</u> organic
radical fuel <u>H</u> electron dependent <u>H</u>	system system <u>W</u> pH cold	burner microscope <u>A</u> quantum kinetic	<u>Y</u> electric subatomic	soda absolute <u>Am</u> unit	particle fossil <u>Et</u> sulfuric
tube charge <u>Gl</u> potential free	balance <u>Ar</u> safety	chemistry zero <u>E</u> metal	fusion earth <u>Lo</u> Nobel cell	conversion acid <u>F</u> sodium carbon	paper scale <u>I</u> ozone
dating motion <u>I</u> ventilation ionic	helix <u>W</u> radioactive	bond energy <u>As</u> teaching Bunsen	reaction gas <u>R</u> metric	decay mass <u>O</u> periodic	number <u>He</u> chain
wash hood <u>St</u> greenhouse noble	principle <u>M</u> test Brownian	<u>O</u> amino point uncertainty	membrane point <u>H</u> balance	acid test <u>S</u> federal grant	variable <u>Li</u> scientific
layer <u>S</u>	<u>T</u> heavy double	solution mechanics <u>T</u> rare surface	<u>K</u> litmus	alloy chloride <u>C</u> atomic eye	lecturer <u>Rd</u> water baking